



BOAT BOSS

Boat Boss Hull Stain Remover and Cleaner

SAFETY DATA SHEET

ISSUE DATE: 8/12/2023 REVISION DATE: 8/12/2023

SECTION 1. PRODUCT IDENTIFICATION

Product Form: Mixture

Product Name: Boat Boss Hull Stain Remover and Cleaner

Intended Use of the Product: Cleaner

Supplier/manufacturers Name: Boat Boss

Address: 221 31st ave N 338, Nashville, TN 37203

Phone: (630)7651473

Website: evoindustrialusa.com

EMERGENCY TELEPHONE NUMBER: 1-800-424-9300 (CHEMTREC)

FOR CHEMICAL EMERGENCY, SPILL, LEAK, EXPOSURE, ACCIDENT, CALL CHEMTREC, 1-800-424-9300

SECTION 2. HAZARDS IDENTIFICATION



Classification (GHS-US) Signal word: Danger

Hazard Statements GHS-US:

EYE: H318

Can cause permanent eye damage. Symptoms include stinging, tearing, and redness and swelling of the eyes. Can cause injury to the cornea.

SKIN: H315

Can cause permanent skin damage. Symptoms may include redness, burning, swelling of the skin, burns, and other skin damage. Passage of this material into the body through the skin is possible but it is unlikely that this would result in harmful effects during safe handling and use.

SWALLOWING: H300

Swallowing this material may be harmful or fatal. Symptoms may include severe stomach and intestinal irritation (nausea, vomiting diarrhea) abdominal pain, and vomiting of blood. Swallowing this material may cause burns and destroy tissue in the mouth, throat and digestive tract. Low blood pressure and shock may occur as a result of severe tissue damage.

INHALATION: H330-H331

Breathing this material may be harmful or fatal. Symptoms may include severe irritation and burns to the nose, throat and respiratory tract.

Precautionary Statements:

P264 - Wash hands, forearms, and exposed areas thoroughly after handling.

P280 - Wear eye protection, protective clothing, protective gloves.

P302+P352 - IF ON SKIN: Wash with plenty of water.

P305+P351+P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P310 - Immediately call a POISON CENTER or doctor.

P321 - Specific treatment (see Section 4).

P332+P313 - If skin irritation occurs: Get medical advice/attention. P362 - Take off contaminated clothing and wash before reuse.

SYMPTOMS OF EXPOSURE:

Kidney damage, stomach, airways.

TARGET ORGANS EFFECTS

Overexposure of this material can cause pre existing disorders, damage to the central nervous system, kidneys.

DEVELOPMENTAL INFORMATION: No Data

CANCER INFORMATION: No Data

OTHER HEALTH EFFECTS: No Data

PRIMARY ROUTES OF ENTRY: Inhalation, Skin Absorption, Skin Contact.

SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredient	Cas # /Name:	% by weight
Oxalic Acid	144-62-7	60-65%
1-Dodecanol, Sulfate, Sodium Salt	151-21-3	35-40%

SECTION 4. FIRST AID MEASURES

Eyes:

Immediately flush eyes with a direct stream of water for at least 15 minutes. Forcibly holding eyelids apart to ensure complete irrigation of all eye and lid tissue. Washing eyes within several seconds is essential to achieve maximum effectiveness. Move individual into fresh air. **GET MEDICAL ATTENTION IMMEDIATELY.**

SKIN

Flush thoroughly with cool water under shower while removing contaminated clothing and shoes. Discard non rubber shoes. Wash clothing before reuse. Wash clothing before reuse. **GET MEDICAL ATTENTION AS SOON AS POSSIBLE.**

INHALATION

Remove to fresh air. If breathing is difficult, have trained person administer oxygen. If respiration stops, give mouth to mouth resuscitation. **GET MEDICAL ATTENTION IMMEDIATELY.**

INGESTION

NEVER GIVE ANYTHING BY MOUTH TO AN UNCONSCIOUS PERSON. If swallowed **DO NOT INDUCE VOMITING.** Give large quantities of water. (If Available, give several glasses of milk) If vomiting occurs spontaneously, keep airway clear and give more water. If possible, do not leave individual unattended. **GET MEDICAL ATTENTION IMMEDIATELY.**

NOTE TO PHYSICIAN

Precipitate oxalate by giving calcium in any form orally, such as milk, lime water, chalk, calcium gluconate, calcium chloride, or calcium lactate. Do not use gastric lavage or emesis if tissue corrosion has occurred. Dissolve 10 g (2 teaspoons) of calcium lactate in lavage or emesis fluids, Antidote give 10% calcium gluconate, 10 ml intravenously and repeat if symptoms persists. Dreisbach, Handbook of poisoning, 11th Edition). Administering antidote and performing lavage should be conducted by medical personnel only. Preexisting disorders of the following organs or organ systems may be aggravated by exposure to this material, skin, lung (for example, asthma like conditions) kidney, central nervous system.

SECTION 5. FIRE FIGHTING MEASURES

Flash Point: Non flammable

Method: Non Flammable

Auto Ignition Temperatures: Non Flammable

Hazardous Products of Combustion: May form carbon dioxide, carbon monoxide, formic acid, acid vapors & sulfur compounds.

EXTINGUISHING MEDIA: Alcohol foam, water fog, carbon dioxide and dry chemical

FIRE FIGHTING PROCEDURES:

Water or foam can cause frothing which can be violent and possibly endanger the live of the firefighter. Water may be used to keep fire exposed containers cool until fire is out. Wear a self contained breathing apparatus with a full face piece operated in the positive pressure demand mode with appropriate turn out gear and chemical resistant personal protective equipment. Refer to the personal protective equipment section of this MSDS.

FIRE & EXPLOSION HAZARD: NO flash to boiling point. Blends which contain a mixture of various hydrocarbons, oxygenated and chlorinated solvents may, due to the volatility of the chlorinated solvents become a combustible or flammable mixture if the volume of chlorinated solvents in the blend is reduced by evaporation during normal usage. The likelihood of this occurring increases as the temperature at which the blend is used increases. Therefore all precautions relating to the use of a combustible or flammable solve should be observed. Oxalic acid crystals are considered to be a combustible solid below 214 degrees F (101 degrees C).

NFPA RATING:

Health:2 Flammability:1 Reactivity: 0

SECTION 6. ACCIDENTAL RELEASE MEASURES

Evacuate unnecessary personnel

SMALL SPILL: Sweep up material for disposal and recovery.

LARGE SPILL: Eliminate all ignition sources (flares, flames including pilot lights, electrical sparks.) Persons not wearing protective equipment should be excluded form area of spill until clean up ha has been completed. Scoop into containers.

SECTION 7. HANDLING AND STORAGE

HANDLING:

Avoid breathing mist. Avoid breathing vapors.

ALWAYS wear ALL protective clothing described above. Containers of this material may be hazardous when emptied. Since emptied containers retain product residues (liquid, vapor and or solid) all hazard precautions given in this data sheet must be observed.

WARNING Sudden release of hot organize chemical vapors or mists from process equipment operated at elevated temperatures and pressure, or sudden ingress of air into vacuum equipment, may result in ignitions without the presence of obvious ignition sources. Published auto ignitions or ignition temperatures values cannot be treated as safe operating temperatures in chemical processes without analysis of the actual process conditions. Any use of this product in elevated temperature processes should be thoroughly elevated to establish and maintain safe operating conditions.

STORAGE:

Keep container tightly closed and properly labeled. Dike storage containers to contain 110% of tank volume. Under normal conditions, this product can be stored satisfactorily in mild steel with out interior lining. Aluminum is not recommended for storage and handling.

SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION



Engineering Controls:

No special ventilation required under normal use. Note: Where carbon monoxide may be generated, special ventilation may be required.

Where engineering controls are not feasible, use adequate local exhaust ventilation where mist, spray or vapor may be generated.

PERSONAL PROTECTION

Respiratory:

Protection is not required under normal use. Wear NIOSHA/MSHA approved respirator following manufacturers recommendations, where airborne contaminants may occur.

Eye/Face

Wear chemical safety goggles, plus full face shield to protect against splashing when appropriate (ANSI z87-1)/

SKIN:

Wear Chemical resistant gloves such as rubber, neoprene or vinyl. Wash contaminated clothing and dry before reuse. Whenever there is a possibility of splash or contact, wear a chemical resistant full body suit and boots.

OTHER:

Standard work clothing closed at the neck and wrists. Discard shoes that cannot be decontaminated. Emergency shower and eyewash facility should be in close proximity (aANSI z358.1)

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

Boiling Point:	200F
Vapor Pressure:	17.500 mmHg
Specific Vapor Pressure:	1.00 @ Air=1
Specific Gravity:	1.02-1.03 @ 77F
Liquid Density:	8.5-8.6 lbs./Gal @ 77F
	1.02-1.03 kg/l @ 25.00 C
% Volatiles:	68-72.0 %
Evaporation Rate:	Slower than Ethyl Ether
Appearance:	Solid
Color:	Pale yellow to white
State:	Solid
Physical Form:	Crystal Solid
Odor:	No Data
pH:	7.8-8.0

SECTION 10. STABILITY AND REACTIVITY

Hazardous polymerizations:

☐ Occurs

☒ Will not occur

Hazardous decomposition:

May form: acid vapors, carbon dioxide, carbon monoxide, sulfur compounds, formic acid.

Chemical stability: Stable

Incompatibility: Avoid contact with:

CHLORINATED, FURFURYL ALCOHOL, HYPOCHLORITES, SILVER, STRONG, ALKALIES,
STRONG OXIDIZING AGENTS,

Acid reacts with most metals to release hydrogen gas which can form explosive mixtures with air.

SECTION 11. TOXICOLOGICAL INFORMATION

NO DATA

SECTION 12. ECOLOGICAL INFORMATION

NO DATA

SECTION 13. DISPOSAL CONSIDERATION

Waste Management Information:

Dispose of in accordance with all applicable local, state and federal regulations.

There is limited information available on the environmental fate and effects of sodium hydroxide (NaOH). Laboratory toxicity data indicate that NaOH is moderately toxic to aquatic and terrestrial organisms. The primary mode of action is due the corrosive nature of this chemical and its tendency to increase pH in poorly buffered environments. Aquatic organisms become increasing stressed as pH exceeds 9, with many aquatic species being intolerant of pH levels in excess of 10. Increased pH due to the introduction of NaOH into aquatic environments may lead to the precipitation of essential micro nutrients. Exposed terrestrial species would be subject to skin irritation and burns due to the corrosive nature of this material, Due caution should be exercised to prevent the accidental release of this material to aquatic or terrestrial environments.

SECTION 14. TRANSPORT INFORMATION

D.O.T. Information

49 CFR 172.101

Corrosive Solid, N.O.S., * UN3261, PG GROUP III

Container Mode:

55 Gal., Drum/ Truck Package

NOS COMPONENT: Oxalic Acid, Dehydrate

REPORTABLE QTY

2000# Ethylene Oxide

SECTION 15. REGULATORY INFORMATION

US FEDERAL REGULATIONS

TSCA STATUS

TSCA (United States) The intentional ingredients of this product are listed.

CERCLA RQ - 40 CFE 302.4 (a)

None Listed.

SARA 302 Components - 40 CFR 355 Appendix A

NONE

SARA 313-COMPONENTS - 40 CFR 372.65

NONE

SARA/TITLE III HAZARD CATEGORIES

Immediate(Acute) Health: X Reactive Hazard

Delayed(Chronic) Health: X Sudden Release of Pressure:

Fire Hazard

HMIS HAZARD RATINGS:

HEALTH HAZARD 2 FIRE HAZARD: 0

REACTIVITY: 2

International Regulations:

DSL (Canada) The intentional Ingredients of this product are listed

State and Local Regulations

California Prop. 65:None

New Jersey RTK Label Information

Oxalic Acid 144-62-7

Pennsylvania RTK Label Information

SECTION 15. ECOLOGICAL INFORMATION

NO DATA

STATE REGULATIONS

See Section 2, COMPOSITION INFORMATION ON INGREDIENTS list legend=d for applicable state regulations.

INTERNATIONAL REGULATIONS:

Consult the regulations of the importing country

CANADA:

WHMIS Hazard Class: D1B, E

SECTION 16. OTHER INFORMATION

IMPORTANT:

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